PALBINONE, A POTENT INHIBITOR OF 3 a - HYDROXY DEHYDROGENASE FROM PAEONIA ALBIFLORA

Shigetoshi Kadota," Satoshi Terashima, Tohru Kikuchi, and Tsuneo Namba

Research Institute for Wakan-Yaku (Oriental Medicines), Toyama Medical and Pharmaceutical University

Summary: The structure of palbinone (1), isolated as potent inhibitor of 3α -hydroxy dehydrogenase from the roots of <u>Paeonia albiflora</u> PALLAS, was determined based on the 2D NMR spectroscopy.

Paeonia Radix, the roots of <u>Paeonia albiflora</u>, is one of the best known crude drugs in Japan, and has been the subjects of many investigators. Especially the monoterpenoids, which are the principal ingredients of Paeonia Radix, have been investigating as a major bioactive component of this crude drug.¹) In our search for biologically significant substances from <u>P. albiflora</u>, we have isolated a new terpenoid, named palbinone (1), and found that it has a potent inhibitory activity against 3α -hydroxy dehydrogenase.²) This paper describes the isolation and structure elucidation of this new terpenoid.

The roots (5.0 kg) of <u>P. albiflora</u> were pulverized and extracted with chloroform at room temperature. The chloroform extract was roughly separated by silica gel column chromatography using CHCl₃ and 5% MeOH/CHCl₃. The CHCl₃ eluates were further separated by preparative TLC to give palbinone (1)(87 mg), along with paeonilactone B.³)

Palbinone (1), red needles (from ether-hexane), mp 254-255°C, $[\alpha]_D -223.8°$ (CHCl₃), showed UV absorptions (MeOH) at 237 and 387 nm (log ε : 3.2 and 3.0)(α , β - γ , δ -unsaturated carbonyl groups) and IR absorptions (CHCl₃) at 3500(OH), 1750(ketone), 1700, 1690(unsaturated ketone), and 1605 cm⁻¹(double bond). It showed the molecular ion peak at m/z 358 in MS and its molecular formula was determined to be C₂₂H₃₀O₄ (M⁺ 358.2137, calcd 358.2143) by high resolution MS. The ¹H- and ¹³C-NMR⁴) and ¹H-¹H and ¹H-¹³C COSY⁵) of 1



255

indicated the presence of two ketones ($\delta_{\rm C}$ 201.26 and 180.89⁶⁾), two double bonds ($\delta_{\rm H}$ 6.40 and 6.90; $\delta_{\rm C}$ 141.57, 120.29, 146.95, and 151.29), a hydroxymethine ($\delta_{\rm H}$ 3.28; $\delta_{\rm C}$ 78.64), five tert-methyl groups ($\delta_{\rm H}$ 0.80, 0.82, 0.93, 1.02, and 1.20; $\delta_{\rm C}$ 15.07, 19.02, 18.32, 27.79, and 19.44), and four quaternary sp³ carbons ($\delta_{\rm C}$ 37.20, 38.96, 40.02, and 50.74). The above data led us to suppose that the structure of palbinone might be 1.

At this stage, we measured the HMBC spectrum⁷⁾ of 1 in order to confirm the assumed structure (1). As shown in formula 2, the 13 C-signals at δ 146.95 (C-13) and at δ 56.08 (C-9) showed long-range correlations with the ¹H-signals at δ 1.20 (30-H₃) and 6.40 (11-H) and δ 0.82 (18-H₃), 0.93 (19-H₃), 6.40 (11-H), and 6.90 (12-H), respectively. In turn, the 13C-signal at δ 37.20 (C-10) was correlated with the ¹H-signals at δ 0.93 (19-H₃) and 2.07 (9-H), and the signal at δ 54.87 (C-5) with the ¹H-signals at δ 0.80 (29-H₃), 0.93 (19-H₃), 1.02 (28-H₃), and 2.07 (9-H). Also some other significant long range correlations are shown by arrows. Thus the planar structure of 1 was proved.

The relative stereochemistry of 1 was determined on the basis of coupling constants of each protons and the results of NOE experiments. Irradiation at the 29-H3 and 18-H3 caused the increase of signal intensity of the 19-, 28-, 6α -, and 6β -protons and the 19-, 6β -, and 7β -protons, respectively, and irradiation at the 19-H₃ and 28-H₃ enhanced the increase of signal intensity of the 29-, 18-, 6β -, and 11-protons and the 29-, 5-, 6α -, and 3-protons, respectively. Also, irradiation at the 30-H₃ gave NOE enhancement of 7α and 9-protons. Therefore the structure of palbinone was proved to be 1.

IC₅₀ value (50% inhibitory concentration) of palbinone (1) against 3α -hydroxy dehydrogenase was 4.6 x 10⁻⁸M. Palbinone (1) was more potent inhibitor of 3α -hydroxy dehydrogenase than indomethacin, which is known as one of the strongest inhibitors with an IC_{50} of 3.2 x $10^{-6}M$ under the same assay conditions.

- 1) M. Kaneda, Y. Iitaka, and S. Shibata, <u>Tctrahedron</u>, <u>28</u>, 4309 (1972); M. Shimizu, T. Hayashi, N. Morita, F. Kiuchi, H. Noguchi, Y. Iitaka, and U. Sankawa, Chem. Pharm. <u>Bull</u>., <u>31</u>, 577 (1983).
- 2) T. M. Penning, J. Pharm. Sci., 74, 651 (1985); T. M. Penning and P. Talalay, Pro.
- Natl. Acad. Sci. USA, 80, 4504 (1983). 3) H. Hayashi, T. Shinbo, M. Shimizu, M. Arisawa, N. Morita, M. Kimura, S. Matsuda, and T. Kikuchi, <u>Tetrahedron Lett</u>., 26, 3699 (1985). 4) 1: ¹H-NMR (CDCl₃) δ : 0.80 (3H, s, 29-H₃), 0.82 (3H, s, 18-H₃), 0.89 (1H, dd, J=12.0, 2.0 Hz, 5-H), 0.93 (3H, s, 19-H₃), 1.02 (3H, s, 28-H₃), 1.13 (1H, td, J=13.0, 4.5 Hz, 1 β -H), 1.20 (3H, s, 30-H₃), 1.51 (1H, ddd, J=14.0, 11.5, 4.0 Hz, 6 β -H), 1.66 (1H, m, 2.2-U) 1.70 (1H, m, 7 α -H) 1.72 (1H, m, 6 α -H), 1.75 (1H, m, 2 α -H), 1.95 (1H, dt. $(2\beta - H)$, 1.70 (1H, m, $7\alpha - H$), 1.72 (1H, m, $6\alpha - H$), 1.75 (1H, m, $2\alpha - H$), 1.95 (1H, dt, J=13.0, 4.0 Hz, 1α -H), 2.06 (1H, m, 7β -H), 2.07 (1H, br s, 9-H), 3.28 (1H, dd, J=12.5, 5.5 Hz, 3-H), 6.40 (1H, dd, J=10.0, 2.5 Hz, 11-H), 6.90 (1H, dd, J=10.0, 3.2 J=12.5, 5.5 HZ, 3-H), 6.40 (1H, dd, J=10.0, 2.5 HZ, 11-H), 6.90 (1H, dd, J=10.0, 3.2 HZ, 12-H); ¹³C-NMR (CDCl₃) δ: 15.07 (29-C), 17.92 (6-C), 18.32 (19-C), 19.02 (18-C), 19.44 (30-C), 26.72 (2-C), 27.79 (28-C), 33.31 (7-C), 37.20 (10-C), 38.11 (1-C), 38.96 (4-C), 40.02 (8-C), 50.74 (14-C), 54.87 (5-C), 56.08 (9-C), 78.64 (3-C), 120.29 (12-C), 141.57 (11-C), 146.95 (13-C), 151.29 (17-C), 180.89 (16-C)⁶), 201.26 (15-C).
 5) A. Bax, "Two-Dimensional NMR in Liquids", D. Reidel Publishing Co., Dordrecht, Holland, 1982; R. Benn and H. Gunther, <u>Angew. Chem. Int. Ed. Engl.</u>, 22, 350 (1983).
 6) E. Pretsch, J. Seibl, W. Simon, and T. Clerc, "Tables of Spectral Data for Structure Determinedian of Compounder" (2120, Springer-Verlag, Parlin, 1982).
- Determination of Organic Compounds", C180, Springer-Verlag, Berlin, 1983.
- 7) A. Bax and M. F. Summers, J. Am. Chem. Soc., 108, 2093 (1986); M. F. Summers, L. G. Marzilli, and A. Bax, *ibid*, *108*, 4285 (1986).

(Received in Japan 2 October 1991)

References and Note